## DEOXYOKAMURALLENE AND ISOOKAMURALLENE, NEW HALOGENATED NONTERPENOID C $_{15}$ -COMPOUNDS FROM THE RED ALGA LAURENCIA OKAMURAI YAMADA $^1)$

Minoru SUZUKI and Etsuro KUROSAWA\*

Department of Chemistry, Faculty of Science,
Hokkaido University, Sapporo 060

Two new  $\mathrm{C}_{15}$ -bromoallenes, deoxyokamurallene and isookamurallene, have been isolated from the title alga, and their structures were determined on the basis of their spectroscopic evidence.

In the recent paper,  $^2$ ) we reported that okamurallene isolated from the red alga Laurencia okamurai Yamada (Rhodomelaceae, collected at Bikuni, Hokkaido) was a unique  $C_{15}$ -metabolite not yet encountered in the halogenated  $C_{15}$ -nonterpenoids from the algae of the genus Laurencia and its structure has been shown by formula k excluding the stereochemistry at C-10 and C-12. In this communication, we describe the structures of two new metabolites, deoxyokamurallene (k) and iso-okamurallene (k), also isolated as the minor constituents from Bikuni's k. okamurai.

Repeated silica gel column and thin-layer chromatography of the neutral extracts ^2) has led to the isolation of deoxyokamurallene (2) (0.2% of the extracts) and isookamurallene (3) (0.4%) along with okamurallene (1) and other unknown  $C_{15}$ -bromoallenes.

Deoxyokamurallene (2),  $C_{15}H_{16}O_2Br_2$  (m/e 390, 388, and 386;  $M^+$ ), oil,  $[\alpha]_D^{23}$ +220° (c 0.84; CHCl $_3$ ), showed in its IR, $^3$ )  $^1$ H and  $^{13}$ C NMR (Table 1) spectra the presence of a bromoallene side chain [ $v_{max}$  (film) 1960 cm $^{-1}$ ;  $\delta$  (CDC1 $_3$ ) 5.45 (1H, dd, J=6.5 and 6.0 Hz) and 6.09 (1H, dd, J=6.0 and 1.5 Hz); δ (CDC1<sub>3</sub>) 202.5 (s), 99.5 (d), and 73.7 (d)], a trisubstituted vinyl ether functionality [ $\nu_{max}$  1660 and 1190 cm<sup>-1</sup>;  $\delta$  158.0 (s) and 90.1 (s)], and a 1,2-disubstituted cyclopropane ring [ $\delta$  17.8 (d), 15.7 (d), and 12.3 (t)] as same as okamurallene ( $\frac{1}{4}$ ). Furthermore, the existence of a -CH=CH-CH $_3$  grouping in  $\stackrel{7}{\sim}$  was indicated by the  $^1$ H NMR spectrum (vide infra). The signals of the olefinic protons were observed at  $\delta$  ca. 5.1 and ca. 5.5 as multiplets, partly overlapping with the signals of the  $C_q\text{-H}$  ( $\delta$  5.04) and the  $C_3$ -H ( $\delta$  5.45), respectively. Irradiation of the vinyl methyl signal at  $\delta$  1.71 (dd, J=7.0 and 1.5 Hz) simplified the multiplet at  $\delta$  ca. 5.5 into a clear doublet (J=11.0 Hz), and the same irradiation effected a change in the shape of the one-proton multiplet due to another olefinic proton at  $\delta$  ca. 5.1. Conversely the methyl signal was collapsed to a doublet (J=7.0 Hz) by irradiation at  $\delta$  ca. 5.1 and further collapsed to a doublet (J=1.5 Hz) by irradiation at  $\delta$  ca. 5.5. The stereochemistry of this double bond is unambiguously indicated to be cis configuration by the value (11.0 Hz) of the coupling constant between the pertinent olefinic protons. Above-mentioned data together with the proton spin decoupling experiments and the close resemblance of the spectral data of 2 and 1 (Table 1) reveal that deoxyokamurallene is 132-13,14-deoxyokamurallene.

Isookamurallene (3),  $C_{15}H_{16}O_3Br_2$  (m/e 406, 404, and 402;  $M^+$ ), oil,  $[\alpha]_D^{27}$  +130° (c 1.00; CHCl<sub>3</sub>), is an isomer of okamurallene (1) and was shown to possess the same carbon skeleton including a bromoallene side chain and a cyclopropane ring as 1 and 2 by comparison of the spectral properties with those of 1 and 2. Further the IR  $[\nu_{max} \ 1715 \ cm^{-1}]$  and  $^1H \ NMR \ [\delta \ 2.17 \ (3H, \ s)$  and 2.56 (2H, d, J=7.0 Hz)] spectra indicate the presence of a -CH-CH<sub>2</sub>-CO-CH<sub>3</sub> moiety in 3, thus showing that isookamurallene must be flanked on C-12 by a -CH<sub>2</sub>-CO-CH<sub>3</sub> grouping instead of -CH-CH-CH<sub>3</sub> in 1.

The stereochemistries at C-10 and C-12 in 1, 2, and 3 were deduced from the

 $^1$ H (100 MHz) and  $^{13}$ C NMR data for okamurallene (L) and deoxyokamurallene (L) in CDC1 $_3$ Table 1.

	Multiplicity, J (Hz)	dd, J=6.0, 1.5		dd, J=6.5, 6.0	ш	dd, J=13.5, 10.0	dd, J=13.5, 5.0			d, J=6.5	dd, J=6.5, 6.5	ш	ш	ш	ш	m (dq, J=11.0, 7.0)	dd, J=7.0, 1.5
8	1 <sub>Н δ</sub>	60.09		5.45	4.35	1.90	2.24			5.30	5.04	1.9	1.0~1.3	1.0~1.3	∿5.1	∿5.5	1.71
	13 <sub>C &amp;</sub> c)	73.7 (d)	202.5 (s)	99.5 (d)	72.1 (d)	40.9 (t)		90.1 (s)	158.0 (s)	88.5 (d)	83.4 (d)	15.7 (d)	12.3 (t)	17.8 (d)	126.3 (d)	128.2 (d)	13.1 (q)
	Multiplicity, J (Hz)	dd, J=6.0, 1.5		dd, J=6.5, 6.0	E	dd, J=13.5, 10.0	dd, J=13.5, 5.0			d, J=6.5	dd, J=6.5, 6.5	E	E	Ħ	dd, J=4.5, 3.0	dq, J=4.5, 5.5	d, J=5.5
<b>-</b> ₽	1н б	6,10		5.44	4.28	1.90	2.24 <sup>b)</sup>			5.31	5.06	∿1.9	1.1~1.3	1.1~1.3	2.80	3.00	1.36
	13 <sub>C 8</sub> a)	73.9 (d)	202.5 (s)	99.2 (d)	72.7 (d)	40.8 (t)		91.2 (s)	157.6 (s)	88.4 (d)	83.6 (d)	12.5 (d)	10.0 (t)	17.0 (d)	52.7 (d)	56.1 (d)	13.7 (q)
	Carbon	1	2	8	4	Ŋ		9	7	∞	6	10	11	12	13	14	15

a) Assignments were made with the aid of proton selective decoupling. b) Previously reported  $^2$  chemical shift of the proton at C-5 should be corrected. c) Assignments were made by comparing the data of 1 and 2.

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С <sub>10</sub> -Н	∿1.9	∿1.9	∿1.9	
$C_{11}^{-H}_{2}$	1.1~1.3	1.0~1.3	0.85, 1.10	
С <sub>12</sub> -Н	1.1~1.3	1.0~1.3	1.45	

Table 2. Chemical shifts of the cyclopropane protons in  $\frac{1}{2}$ , and  $\frac{3}{2}$ 

chemical shifts of the cyclopropane protons (Table 2). When the substituent at C-12 changed from -CH-CH-CH<sub>3</sub> in 1 into -CH=CH-CH<sub>3</sub> in 1, no marked effect on the chemical shifts of the 10-H, the 11-H<sub>2</sub>, and the 11-H was observed. On the other hand, in 13, the remarkable change in the chemical shifts of the 11-H and one of the methylene protons at C-11, which seems to be 12-to the substituent at C-12, was observed and 13-H signal showed no prominent shift from that of 13, suggesting that the proton at C-10 should be 12-to the substituent at C-12.

Thus the structures of okamurallene, deoxyokamurallene, and isookamurallene would be represented by formulas 1, 2, and 3 respectively.

## References

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- 2) M. Suzuki and E. Kurosawa, Tetrahedron Letters, 22, 3853 (1981).
- 3) IR data for deoxyokamurallene (2):  $v_{\rm max}$  (film) 3040, 1960, 1660, 1190, 1115, 1075, 1025, 935, 910, and 850 cm<sup>-1</sup>.
- 4) IR,  ${}^{1}$ H and  ${}^{13}$ C NMR data for isookamurallene (3):  $\nu_{max}$  (film) 3050, 1965, 1715, 1660, 1200, 1118, 1035, 1015, 995, 860, and 845 cm<sup>-1</sup>;  $\delta$  (100 MHz, CDC1<sub>3</sub>) 0.85 (1H, m; C-11), 1.10 (1H, m; C-11), 1.45 (1H, m; C-12), 1.8-2.1 (2H, m; C-5 and C-10), 2.17 (3H, s; C-15), 2.24 (1H, dd, J=13.5 and 5.0 Hz; C-5), 2.56 (2H, d, J=7.0 Hz; C-13), 4.30 (1H, m; C-4), 5.04 (1H, dd, J=6.5 and 6.5 Hz; C-9), 5.29 (1H, d, J=6.5 Hz; C-8), 5.45 (1H, dd, J=6.5 and 6.0 Hz; C-3), and 6.10 (1H, dd, J=6.0 and 1.5 Hz; C-1);  $\delta$  (CDC1<sub>3</sub>), 207.5 (s; C-14), 202.5 (s; C-2), 158.0 (s; C-7), 99.4 (d; C-3), 91.1 (s; C-6), 88.4 (d; C-8), 83.6 (d; C-9), 74.0 (d; C-1), 72.5 (d; C-4), 43.4 (t; C-13), 40.9 (t; C-5), 29.6 (q; C-15), 14.5 (d; C-12), 12.8 (d; C-10), and 11.0 (t; C-11).

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